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Synthesis, Characterization and Visible light/NIR Photodetector of CuO Nanowires Fabrication

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Abstract. Successfully synthesized vertically aligned, high density, long, large aspect ratio of CuO nanowires (CuO NWs) via thermal oxidation of Cu foil with atmospheric air. The oxidation process was conducted under temperature and time variation. The morphology and structure properties of CuO NWs were investigated by scanning electron microscope (SEM) and X-ray diffraction (XRD) respectively. UV-Visible spectrophotometer was also used to estimate the direct band gap (1.75eV) from the transmission spectrum. Furthermore, the film exhibit high absorption in the visible range and near infrared wavelength. The optimized parameters for long, small diameter, high density of CuO NWs at oxidation temperature of 500°C for 6-8h in the atmospheric air were found. The diameter and length of CuO NWs was determined within the range of 50-100 nm and 20-30 µm respectively. These optimized properties of CuO NWs make it a dependable candidate for Visible light / Near-Infrared photodetector fabrication. Therefore, the photodetector measurements were proved to render an acceptable performance.

Keywords: Thermal oxidation, Copper oxide nanowires (CuO NWs), Visible light/NIR, Photodetectors.

1. Introduction

One-dimensional metal-oxide semiconductor nanostructures, such as whiskers and nanowires, have attracted increasing interest, due to their unique properties and variety of potential applications [1-4]. Among the metal oxides, cupric oxide (CuO) has been extensively studied as a p-type semiconductor. It has a direct band gap in the range of 1.2 to 2.0 eV and exhibits many excellent chemical and physical properties [5, 6]. After the first discovery of CuO whiskers [7], various growth techniques have been developed to synthesize CuO nanostructures, such as hydrothermal method [8], wet chemical method [9], thermal oxidation [10,11] and the template based sol-gel route [12]. Among these techniques, thermal oxidation is interesting because it is a simple, cheap and produces high quality NWs. Many researchers have reported on the growth of CuO nanowires (CuO NWs) by a thermal oxidation technique [13.14]. Various structures and shapes of copper oxide nanostructures have been investigated which includes: nanowhiskers [15], nanowires [16], flower-like [17], nanoneedles [18], nanorods [19], nanotubes [20], nanocrystals [21], nanosheets [22], nanoribbons [23]. Nanostructured CuO materials, especially 1D CuO NWs, have received much more attention for some application such as field-effect transistors [24], gas sensors [25], UV light sensing [26], visible light photodetector [27], IR photodetector



[28,29], UV/IR photodetector [30,31]. Other materials like CdS NPs dopted with copper and their analysis as p-Cds /n-ZnO synthesized as a thin film photodiode has been used in a photodiode application [32]. Furthermore, metal oxide composite was also fabricated as a photodiodes [33]. Graphene-organic semiconductor hybrid were also fabricated as materials for new photodiodes device [34]. In this work, we present the successful growth of vertically aligned CuO NWs of copper foil thermally oxidize in atmospheric air using tube furnace. The results show an investigation of surface morphologies, crystal structures of CuO NWs produced at different temperatures of oxidation and different times. The optimum growth temperature of vertically aligned CuO NWs produced via thermal annealing of copper foil was determined. Besides the synthesis and characterization of CuO NWs, their performance as a photodetector fabrication for visible light / NIR was also investigated.

2. Experimental details

2.1 Thermal oxidation

Before the CuO nanowires (CuO NWs) growth, pieces of commercial grade copper foil (20mmx20mmx0.2mm) were washed with soap solution. The foils were washed again with dilute hydrochloric acid to remove the native oxide layer, then rinsed in deionized water and followed by cleaning with acetone, ethanol, and deionized water under an ultrasonic bath for 15 min and, finally, dried with nitrogen (N_2) flow. The thermal oxidation process of the foil was conducted in a single zone tube furnace with quartz tube (5 cm diameter, 100cm long). The cleaned copper foils were loaded on alumina boat for each step of temperature and inserted into the middle of the quartz tube. The tube furnace was heated to the set-point temperature of 450°C, 500°C, and 550°C at atmospheric pressure and with time variation. The furnace then switched off and allowed to cool naturally to room temperature to prevent the film from cracking by thermal stress by air. The sample was pulled out of the furnace for further characterization.

2.2 Characterizations

X-ray diffraction data of the CuO powder were obtained by using a Shimadzu XRD-6000 X-ray diffractometer working with Cu K α radiation ($\lambda = 1.5406$ Å) and scanned with speed of 12° min⁻¹ and the diffraction angle ranges from 20° to 80°. The morphology of the films was imaged with Scanning Electron Microscope (Inspect S50, FEI Company, Netherland). UV-Visible spectrophotometer (UV-Vis T80 /PG Instrument Ltd) was used for optical properties calculation.

3. Results and Discussion

3.1 Morphological analysis

The CuO NWs morphology was investigated at different oxidation temperatures of 450°C, 500°C, 550°C and time of 2-4h. Figure 1 shows the SEM images of CuO NWs growth process monitoring. The nanowires started to grow with minor distributed nanowires at a



temperature of 450°C for 2h as shown in Figure 1a. Figure 1b shows the nanowires growth increased at the same temperature for 4h. When the temperature is increased to 500°C for 2 and 4h, the nanowire was grown with increasing density and length as shown in Fig's 1b and c respectively. As the temperature increased to 550°C for 2h and 4h, the lower density of nanowires growth was found as shown in Fig.1 e and f respectively.

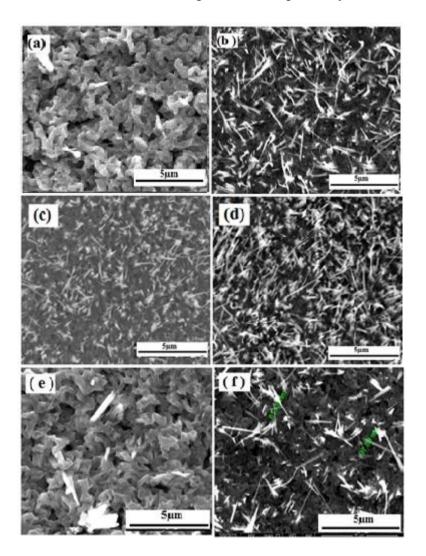


Fig. 1 SEM images of CuO NWs grown at various temperatures and times (a,b) 450°C,2h,4h; (c,d) 500°C,2h,4h; (e,f) 550°C, 2h,4h.

Figure. 2 shows the top view (a and c) and tilted 40° (b and d) of SEM images for the typical as-synthesized CuO nanowires. Figures 2b and 2d images are revealed a long, thin, and vertically aligned on the substrate surface. The measured sizes of these wires are between 100 and 150 nm in diameter and its lengthis greater than 10µm. Two different diameters can be observed along the wires. The bottom half diameter is the vertically aligned wires which were previously measured. Thesecond top half is the long, uniform, crossing over and finediameters of the nanowires, which could be having a large aspect ratio.



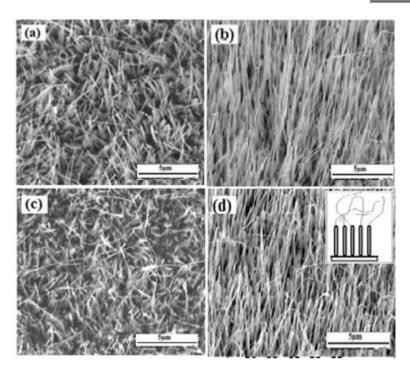


Fig. 2 SEM images of as-synthesized CuO NWs grown at constant temperature and various times. (a,b) 500°C, 6h; (c,d) 500°C, 8h respectively.

The benefit of this work is to obtain CuO NWs with diameters range of 50-100 nm and 10-20 μ m long (large aspect ratio about 500). Vertical CuO nanowires were grown in a simple oxidation process. This product could be controllable and scalable for some application. The optimum temperature and time are 500°C and 6-8h respectively to grow vertically aligned CuO NWs from copper foil were found.

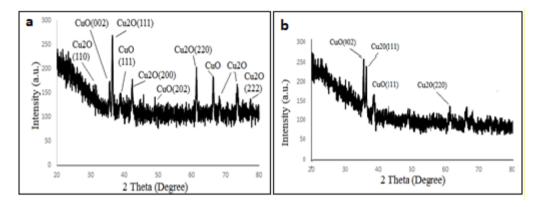
3.2 Structural analysis

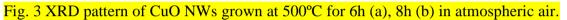
Figure 3 shows a typical XRD pattern of the copper foil was oxidized at 500°C for 6h and 8h in atmospheric air, respectively. It can be seen that there are two phases of copper oxide in both figures. These phases are indexed as CuO and Cu₂O relative to the PDF files No. 05-0661 and 05-666 respectively. Figure 3a shows the XRD pattern of copper foil oxidized at 500°C for 6 hours. In this figure, Cu₂O peaks are the dominant compared to the CuO peaks. This is because, during the Cu foil oxidation, Cu₂O film was firstly formed before the growing of the CuO NWs. Figure 3b shows the XRD pattern of copper foil oxidized at 500°C for 8 hours.Mixed phases are also observed in this figure for CuO and Cu₂O with higher intensity of CuO.Therefore, it is possible to fabricate visible light photodetector which based on the preferred and dominated CuO NWs structure shown in Fig.3b.

The mechanism of CuO NWs grown by an oxidation process of Cu foil is explained in three steps. In the 1^{st} step, the existing of the hillocks to relieve the compressive stress created at the Cu foil at a certain temperature and time in the air. In the 2^{nd} step, Cu₂O is formed by the oxidation of the top surface in air ambient. In the 3^{rd} step, the transformation of the Cu₂O to



the CuO NW. Therefore, the main sources of CuO NW growth are the continuous feeding of Cu from the copper foil and O_2 from the ambient air.





3.3 Optical properties of CuO NWs

Generally, the optical properties of the film studied depend on many factors like crystal structure, incident photon energy, film thickness, and film surface morphology [28]. The optical gap of CuO NWs was calculated using Eq. 1 [35]:

$$(hv) = \beta (hv - Eg)^n / \alpha \qquad (1)$$

Where hv is the incident photon energy, α is the absorption coefficient, β is a material dependent constant and Eg is the optical energy gap. The value of n depends on the nature of the transition. This value takes 1/2, 3/2, 2 or 3 for direct allowed, direct forbidden, indirect allowed indirect forbidden transitions respectively[36]. The usual method of determining Eg involves plotting (α hv)^{1/n} versus hv.

Figure 4 (a and b) shows the relation of CuO NWs transmittance versus the wavelength, synthesized at 500°C for 6h and 8h in atmospheric air. The average transmittance of both curves is found around 60-80% in visible and IR regions (500nm-1100nm) and higher. The insets in Fig. 4a and b display the relation between $(\alpha h v)^2$ and the photon energy (hv) for which its oxidized at 500°C for 6h and 8h in atmospheric air, respectively. The direct energy gap values of 2.1 and 1.6 eV for Cu₂O and CuO NWs were calculated respectively. These low band gap values are exhibited a broadband photoresponse in the visible / IR region. The obtainedband gap values for Cu₂O and CuO are within the ranges of (2.10 eV - 2.61 eV) and (1.20-2.10eV)as reported [37,38] respectively. These optical properties are in a good agreement compared with other reported results [28, 39].



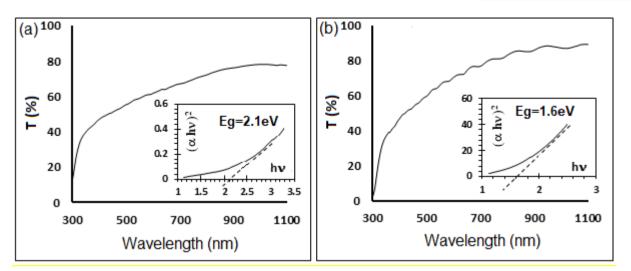


Fig. 4 (a) Transmittance for CuO NWs synthesized at 500°C for 6h(a) and 8h (b). The insets are represent a corresponding energy gap calculations.

4. CuO NWs based visible light photodetector fabrication

For the fabrication of the photodetector, the sample as prepared was peeled from the copper metal by fine razor and fixed on a glass slide. Silver paste was used to form solid contact pads on the top of the detector at both ends of the film only. This was left in the oven to dry at 100°C for 30min. Figure 5a, b, and c demonstrate the photodetector device assembly. Then the detector device was placed on an insulated sheet in the ambient atmosphere. The light power was varied using a variable transformer (AC) and measured with the power meter (Sanwa /Mobiken-Laser power meter LP1). Other instruments such as DC power supply (LONG WEI PS-305DM-China), precise digital timer (Sports timer /Germany), 100watt white light source (Halogen) and manual light switch were used for experimental measurements. The optical power supplied to the light source was varied through variable resistance (Variac). The experimental data were collected with Lap Top connected and pre-installed to Fluke 8846A Digital Multimeter. Figure 5a shows the schematic of the detector device. The sample was fixed under pre-calibrated visible light source as shown in Fig. 5 (c) was calculated approximately 0.24cm².

The current-voltage characteristics of the photodetector device for CuO NWs on glass were measured in the dark and at ambient atmosphere is shown in Fig. 5d. The linear curve is observed with a high slope of 5 M Ω . There is a minor nonlinear behavior was observed around 0 volts (< 0.5v) as explained in the inset of Fig.5d. This behavior is due to the Schottky barrier between CuO NWs and the silver electrode. The outcomes indicate that an ohmic contact is established between the silver electrode and the Cu wire.



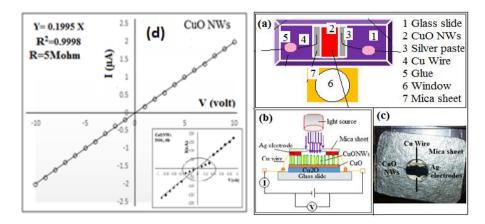


Fig.5 Photodetector of CuO NWs (a) Schematic of the device assembly; (b) Setup measurement diagram; (c) Photograph of the detector device; (d). Current-voltage (I-V) curve in the dark.

4.2 Photodetector performance

In the present work, CuO NWs on the slide glass substrate was studied as a photo-detector at room temperature and atmospheric air. The photoresponse of the CuO NWs detector proved to have a good response to visible light. Figure 6a shows the I-V curves of the device under visible light irradiation with the various light intensity which it is decreased from 166mW/cm² to 0.3mW/cm². The I-V curves also, reveals an ohmicbehavior with good linearity. In addition, we can see that, as the incident light intensity decreased, the maximum current value also decreased from 9µA and lower at the bias voltage of 5 volts. The photocurrent within1µA was measured in the dark at 5volt and so, the calculated sensitivity range of this detector [S (%) = (I_{phot}-I_{dark}) / I dark] is varied from 100-800, which is quite acceptable compared with other workers results [40]. The photocurrent (I) versus light intensity has been described as a power law [41]:

$$\mathbf{I} = \mathbf{A} * P^b \qquad (2)$$

Where I is the photocurrent; A is proportionality constant; P is the light intensity, and b is the empirical value. Figure 6b shows the relation between the photocurrent I (μ A) as a function of light intensity P (mW/cm²). By fitting the measured data (Experimental-solid dark line) with the power law, we obtain the intensity law, I =1.4*P ^{0.34}, where I is the photocurrent and P is the light intensity. The none unity exponent (b=0.34) is a consequence of the complex process of (electron-hole) generation, trapping and recombination within the semiconductor [42].



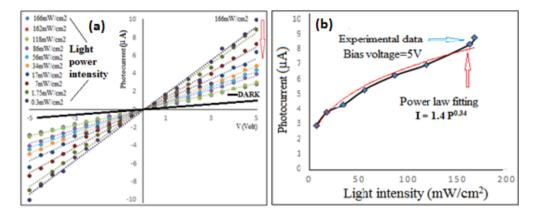


Fig.6 (a) I-V curves for the detector exposed to the light at various power densities (Black solid line represents: dark current); (b) Light intensity dependence of the photocurrent measured at 5 v bias voltage (black solid line). The red solid line is the fitting results.

Figure 7 explains the time dependence photoresponse of this device measured under the fixed visible light intensity of 166mW/cm² at various voltages (1-5volt). These curves are evaluated by manual ON/OFF cycles.Figure7a represent the photocurrent variation following the incident light switching between ON and OFF cycles. This indicating that the detectable response is acceptable for this device. Moreover, an excellent device performance was confirmed by their stability and reversibility at the measured ON/OFF cycles. Moreover, the curves in Fig.7b explain the time dependence photoresponse at the fixed bias voltage of 5 volts and variable visible light intensity ranging from 166mW/cm² to 42mW/cm². Fig. 7c shows clearly the time dependence photoresponse at fixed values of a bias voltage of 5 volts and power intensity of 166mW/cm². Figure 7d shows ON/OFF for two cycles, where the rise and decay time are counted within the 40s and 50s respectively, which are compared with other workers results [43].



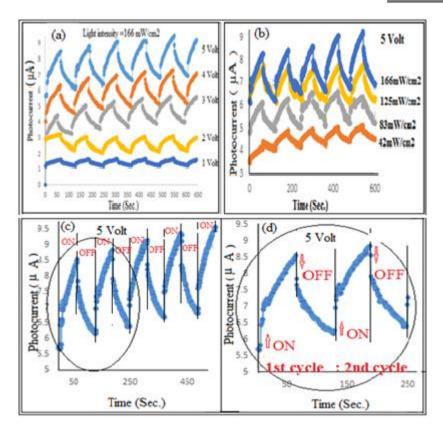


Fig.7 Time-dependent photocurrent response for CuO NWs; (a) at various voltages and fixed light intensity of 166mW/cm²; (b) at various light intensity and fixed bias voltage of 5 volts; (c and d) Multiple and double photocurrent response cycles at the light intensity of 166

 mW/cm^2 and biased voltage of 5volt respectively.

4.3 Photodetector evaluation

There are two important parameters which are spectral responsivity (R_{λ}) and quantum efficiency (η) for the photodetector can determine the sensitivity for optoelectronic applications. They are described in Eq. 3 and Eq. 4 [44]:

 $R_{\lambda} = \Delta I / PA$ (3)

 $\eta = hcR_{\lambda}/e\lambda(4)$

Where $\Delta I (\Delta I = I_{photo} - I_{dark})$ is the photocurrent and the dark current difference, P is the light power intensity irradiated on the CuO NWs, A is the effective area of the irradiated sample, h is the photocurrent, c is the velocity of light, h is the Planck's constant, e is the electronic charge, and λ is the exciting wavelength.

Figure 8 a and b are explaining the relation between the spectral responsivity and the quantum efficiency as a function of the wavelength of constant bias voltage (5Volt)of the CuO NWs prepared by thermal oxidation of copper sheet. Regarding these results, the device exhibits high responsivity (0.25 A/W) for visible at 500nm and 0.39 A/W for IR



light at 1000nm as shown in Fig. 8a. The quantum efficiency (η) is an important evaluation parameter for photodetector performance. It is related to the number of electron-hole pairs excited by absorbed photons [28]. The quantum efficiency values of CuO NWs photodetector (Fig. 8b) are 62% (at 500nm) and 48% (at 1000nm). These values suggest that the synthesized CuO NWs exhibited a high detection performance to visible light and IR.

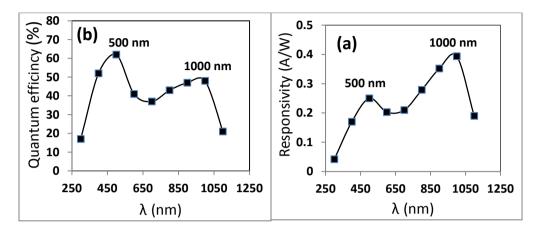


Fig.8 Calculated the photodetector parameters for CuONWs at visible to IR (a) responsivity (R); (b) quantum efficiency (η %) vs. wavelengths.

5 Conclusions

In summary, we have successfully synthesized CuO NWs via a thermal oxidation method at various temperatures and time. Optimized growth parameters to grow CuONWs were determined. The optimized temperature and time values to grow nanowires were 500°C and 6-8h respectively. The diameter of the length of CuO NWs was determined within the range of 50-100 nm and 20-30 µm respectively. The high aspect ratio (L/D) of nanowires could be a serious benefit for many applications. The as-synthesized CuO NWs was revealed high sensitivity to visible light /IR irradiation in the wavelength range of the 400nm-1000nm. A CuO NWs photodetector device was fabricated and this device exhibits a reasonable sensitivity. I-V curves under dark and light response were investigated. Upon visible light illumination, the detector exhibits high sensitivity at 400-1000nm with different bias voltages at room temperature and in an air atmosphere. We thus conclude that our CuO NWs photocurrent detector is quite promising as a visible / NIR light detector fabricated for some applications.

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